2004 Vol. 6, No. 6 1033–1036

## Appending a Tris-imidazole Ligand with a Tyr<sup>244</sup> Mimic on the Distal Face of Bromoacetamidoporphyrin

James P. Collman,\* Richard A. Decréau, and Simona Costanzo

Department of Chemistry, Stanford University, Stanford, California 94305-5080 jpc@stanford.edu

Received January 14, 2004

## **ABSTRACT**

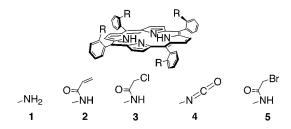
Bromoacetamidoporphyrin is a convenient synthon for the attachment of distal superstructures at room temperature in good yields. New models are presented that contain a tris-imidazole distal ligand set bound to the porphyrin in either a binary or trinary fashion. More importantly, one distal imidazole is cross-linked to a phenol mimicking  $Tyr^{244}$ , making this model the closest structural analogue yet reported of the metal free cytochrome c oxidase (CcO) active site.

Structural characterization of cytochrome c oxidase (CcO) active sites  $^{1-2a-c}$  has led to significant progress in the design and synthesis of close structural CcO analogues. Several linkers have been attached to amino-porphyrin 1 to perform the tethering of the distal superstructure:  $^{3a-c}$  Michael acceptor porphyrins (2),  $^{3d-e}$  chloroacetamidoporphyrins (3),  $^{3e-g}$  and isocyanatoporphyrins (4).  $^{3h}$  Some, such as 4, are moisturesensitive or have been used with moisture-sensitive compounds such as the acyl chloride superstructures in reaction

with 1. On the other hand, 2 and 3 are very stable in air, but heating is required for them to react, which induces significant rotation, lowering the yields and preventing recovery of the unreacted starting material. A new derivatization of porphyrin anilines is presented to obtain bromoacetamidoporphyrin synthon 5 offering a compromise between reactivity and moisture sensitivity.

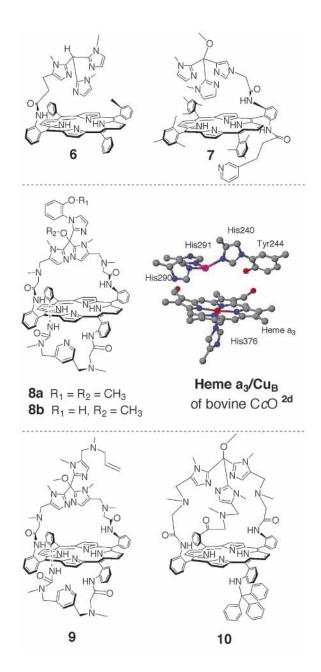
(3) (a) Collman, J. P.; Gagne, R. R.; Reed, C. A.; Halbert, T. R.; Lang, G.; Robinson, W. T. J. Am. Chem. Soc. 1975, 97, 1427. (b) Collman, J. P.; Broring, M.; Fu, L.; Rapta, M.; Schwenniger, R. J. Org. Chem. 1998, 63, 8084. (c) Collman, J. P.; Sunderland, C.; Boulatov, R. Inorg. Chem. 2002, 41, 2282. (d) Collman, J. P.; Zhang, X.; Herrmann, P. C.; Uffelman, E. S.; Boitrel, B.; Straumanis, A.; Brauman, J. I. J. Am. Chem. Soc. 1994, 116, 2681. (e) Collman, J. P.; Boitrel, B.; Fu, L.; Galanter, J.; Straumanis, A.; Rapta, M. J. Org. Chem. 1997, 62, 2308. (f) Collman, J. P.; Broring, M.; Fu, L.; Rapta, M.; Schwenniger, R.; Straumanis, A. J. Org. Chem. 1998, 63, 8082. (g) Lindsey, J. S.; Prathapan, S.; Johnson, T. E.; Wagner, R. W. Tetrahedron 1994, 50, 8941. (h) Collman, J. P.; Wang, Z.; Straumanis, A. J. Org. Chem. 1998, 63, 2424.

<sup>(1)</sup> Ferguson-Miller, S.; Babcock, G. T. Chem. Rev. 1996, 96, 2889. (2) (a)Iwata, S.; Ostermeier, C.; Ludwig, B.; Michel, H. S. Nature 1995, 376, 660. (b) Tsukihara, T.; Aoyama, H.; Yamashita, E.; Tomizaki, T.; Yamaguchi, H.; ShinzawaItoh, K.; Nakashima, R.; Yaono, R.; Yoshikawa, S. Science 1996, 272, 1136. (c) Tsukihara, T.; Aoyama, H.; Yamashita, E.; Tomizaki, T.; Yamagushi, H.; Shinzawaitoh, K.; Nakashima, R.; Yaono, R.; Yoshikawa, S. Science 1995, 269, 1069. (d) Yoshikawa, S.; Shinzawa-Itoh, K.; Nakashima, R.; Yaono, R.; Yamashita, E.; Inoue, N.; Yao, M.; Fei, M. J.; Libeu, C. P.; Mizushima, T.; Yamagushi, H.; Tomizaki, T.; Tsukihara, T. Science 1998, 280, 1723.



Since the stability of the Cu complex in the tris-imidazole pocket might be an issue,4 and tris-imidazolyl methane gives stable complexes with copper(I) ions,<sup>5</sup> Naruta et al.<sup>6a</sup> and Collman et al.6b synthesized a new family of active-site models with 6 and 7, in which a tris-imidazolyl moiety is covalently linked to the porphyrin. However, this linkage to the porphyrin was attached at only one position, which allowed the triimidazole ligand to rotate. To obviate this problem, a new family of cytochrome c oxidase models has been developed (models 8-10). In models 8 and 9, the distal tris-imidazole moiety and the proximal base are cross-translinked to the  $\alpha\beta\alpha\beta$ -porphyrin atropisomer in a binary fashion reminiscent of previous porphyrin designs, 7a-d whereas in model 10, the former is linked in a trinary fashion to the  $\alpha\alpha\alpha\beta$ -porphyrin. A phenol moiety that our previous models lacked has been covalently linked to one distal imidazole in model **8b**. It mimics the Tyr<sup>244</sup> residue, which is thought to play a key role in the  $4H^+/4e^-$  reduction of  $O_2^{8a-e}$  and which has been the subject of studies based on non-heme models. 8f-j Model **8b** is a more representative model of the CcO active site because all key groups are present: porphyrin, proximal base, three distal imidazoles, and one imidazole linked to a phenol residue. In model 9, an amine remains protected and after deprotection may be used as a synthon for modifying the superstructure.

Several synthetic routes were investigated for the preparation of the pyridine-diamine building block **16** (Scheme 1). Following a previously reported procedure<sup>9a</sup> with some modifications, 3,5-pyridine-dicarboxylic acid was converted into the diacyl chloride **17**, which was further reduced by



lithium tri-*tert*-butoxyaluminum hydride to afford the dial-dehyde **18** in 28% yield. <sup>9a</sup> To avoid the use of hydride in the preparation of **18**, **17** was converted into the pyridine-bis-3,5(p-toluenesulfonylhydrazide) **19** in 82% yield.

This underwent a McFayden—Stevens reaction by treatment with Na<sub>2</sub>CO<sub>3</sub> to afford **18** in 25% yield. A simpler approach employed a Riley oxidation of 3,5-dimethyl pyridine yielding **18** in 20% yield. Pyridine dicarbinol **20** was oxidized to **18** in 32% yield by treatment with MnO<sub>2</sub>. LiAlH<sub>4</sub> reduction of pyridine diester **21a,b** led to **20** in poor yields, contrary to previous reports. <sup>9b-e,7c</sup> However, the polymer-supported borohydride reduction of **17**, previously reported by Ley for the reduction of pyridine acyl chloride species, <sup>9d</sup>

1034 Org. Lett., Vol. 6, No. 6, 2004

<sup>(4)</sup> Collman, J. P.; Boulatov, R.; Sunderland, C. J. In *The Porphyrin Handbook*; Academic Press: Boston, 2003; Vol. 11, Chapter 63, pp 1–49. (5) Sorrell, T. N.; Borovik, A. S. *J. Am. Chem. Soc.* **1987**, *109*, 4255.

<sup>(6) (</sup>a) Tani, F.; Matsumoto, Y.; Tachi, Y.; Sasaki, T.; Naruta, Y. Chem. Commun. 1998, 1731. (b) Collman, J. P.; Zhong, M.; Wang, Z.; Rapta, M. Org. Lett. 1999, 1, 2121.

<sup>(7) (</sup>a) Boitrel, B.; Lecas, A.; Renko, Z.; Rose, E. New J. Chem. 1989, 13, 73. (b) Momenteau, M.; Mispelter, J.; Loock, B.; Lhoste, J.-M. J. Chem. Soc., Perkin Trans. 1 1985, 61. (c) Momenteau, M.; Mispelter, J.; Loock, B.; Bisagni, E. J. Chem. Soc., Perkin Trans. 1 1983, 189. (d) Maillard, P.; Schaeffer, C.; Huel, C.; Lhoste, J. M.; Momenteau, M. J. Chem. Soc., Perkin Trans. 1 1988, 3285

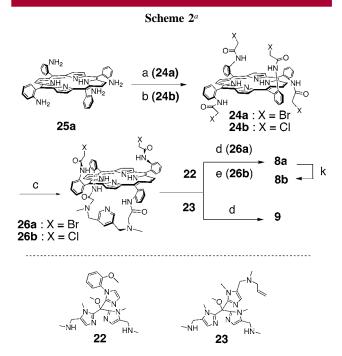
<sup>(8) (</sup>a) Gennis, R. B. Biophys. Biochim. Acta 1998, 1365, 241. (b) MacMillan, F.; Kannt, A.; Behr, J.; Prisner, T.; Michel, H. Biochemistry 1999, 38, 9179. (c) Proshlyakov, D. A.; Pressler, M. A.; Babcock, G. T. Proc. Natl. Acad. Sci. U.S.A. 1998, 95, 8020. (d) Proshlyakov, D. A.; Pressler, M. A.; DeMaso, C.; Leykam, J. F.; DeWitt, D. L.; Babcock, G. T. Science 2000, 290, 1588. (e) Sucheta, A.; Szundi, I.; Einarsdottir, O. Biochemistry 1998, 37, 17905. (f) Collman, J. P.; Zhong, M.; Constanzo, S.; Zhang, C. J. Org. Chem. 2001, 66, 8252. (g) Collman, J. P.; Wang, Z.; Zhong, M.; Zeng, L. J. Chem. Soc., Perkin Trans. 1 2000, 8, 1217. (h) Aki, M.; Ogura, T.; Naruta, Y.; Le, T. H.; Sato, T.; Kitagawa, T. J. Phys. Chem. A 2002, 106, 3436. (i) Cappuccio, J. A.; Ayala, I.; Elliott, G. I.; Szundi, I.; Lewis, J.; Konopelski, J. P.; Barry, B. A.; Einarsdottir, O. J. Am. Chem. Soc. 2002, 124, 1750. (j) Kamaraj, K.; Kim, E.; Galliker, B.; Zakharov, L. N.; Rheingold, A. L.; Zuberbuhler, A.; Karlin, K. D. J. Am. Chem. Soc. 2003, 125, 6028.

<sup>(9) (</sup>a) Choma, C. T.; Kaeslte, K.; Akerfeldt, K. S.; Kim, R. M.; Groves, J. T.; DeGrado, W. F. *Tetrahedron Lett.* **1994**, *35*, 6191. (b) Queguiner, G.; Pastour, P. *Bull. Soc. Chim. Fr.* **1968**, 4117. (c) Palecek, J.; Ptackova, L.; Kuthan, J. *Coll. Czechosl. Chem. Commun.* **1969**, *34*, 427. (d) Habermann, J.; Ley, S. V.; Scott, J. S. *J. Chem. Soc.*, *Perkin Trans. 1* **1999**, 1253.

<sup>a</sup> Reagents and conditions: (a) SOCl<sub>2</sub>, DMF cat., toluene, 110 °C, 95%; (b) (*t*-BuO)<sub>3</sub>AlH, THF, −78 → 0 °C, 28%; (c) TsNHNH<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>/C<sub>6</sub>H<sub>5</sub>N, 40 °C, 82%; (d) SeO<sub>2</sub>, dioxane—water, 80 °C, 12 h, 20%; (e) polymer—N<sup>+</sup>Me<sub>3</sub>BH<sub>4</sub><sup>-</sup>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 30 min, 43%; (f) MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt, 2 h, 32%; (g) K<sub>2</sub>CO<sub>3</sub>, glycol, 150 °C, 5 min, 25%; (h) LiAlH<sub>4</sub>, THF, −78 → 0 °C, 5−10%; (i) CH<sub>3</sub>NH<sub>2</sub>, CH<sub>3</sub>OH, rt, 16 h, 50%; (j) NaBH<sub>4</sub>, CH<sub>3</sub>OH, 10 min, 50%; (k) ROH, H<sub>2</sub>SO<sub>4</sub>, 80 °C, 3 h, 80%.

yielded **20** in 43% yield. Diimination of **18** by treatment in methylamine followed by reduction with NaBH<sub>4</sub> afforded the desired diamine pyridine **16** in 65% yield.

Preparation of  $\alpha\beta\alpha\beta$ -models 8 and 9 was achieved by  $S_N2$ reaction of bis-secondary amine straps 16, 22, and 23 with a tetrahalogenoacetamidoporphyrin 24a,b (Scheme 2). The sequence was chosen to introduce the pyridine tail 16 first, followed by the tris-imidazole ligand 22 or 23 because the purification of imidazole-containing porphyrins is always tedious. The previously reported chloroacetamidoporphyrin TCIPP 3 <sup>3e,f</sup> required halogen exchange and heating (45-60 °C) for 18-72 h to be condensed with secondary amines in ca. 40% yield. 3c But a control reaction showed that  $\alpha\beta\alpha\beta$ -TCIPP **25b** refluxed in acetone (55 °C) for 6 h led to 50% rotation. Therefore, the preparation of bromoacetamido-type porphyrins **24a**,**c**,**e** was carried out since Br is a better leaving group for S<sub>N</sub>2 reactions and should allow milder reactions. Tetra(bromoacetamido)-porphyrin **24a** ( $\alpha\beta\alpha\beta$ -TBrPP) is obtained in 86% yield by addition of bromoacetyl bromide to  $\alpha\beta\alpha\beta$ -tetraaminophenyl-porphyrin synthon **25a**. A short reaction time and low temperature are required; otherwise, side-products form. Preparation from a pure TAPP-atropisomer was preferred over a mixture of TAPP-atropisomer because separation of TBrPP-atropisomers have closer  $R_f$ values than TAPP, making them more difficult to separate. Subsequent reaction of 24a and diamine 16 proceeded at room temperature, affording 26 in 45% yield, with no observed rotation. The same reaction run with TClPP 24b led to 14% of **26b** after a reaction time of several weeks at room temperature and 23% yield when the reaction was carried out at 50 °C. The subsequent reaction of 26a with tris-imidazole dipodal ligands 22% and 23 proceeded at room temperature and led to 8 and 9 in 40 and 35% yields, respectively. Longer reaction times led to a drop in yield of 8 and 9 together with other unidentified fractions, presumably arising from S<sub>N</sub>2 nucleophilic attack of the unprotected



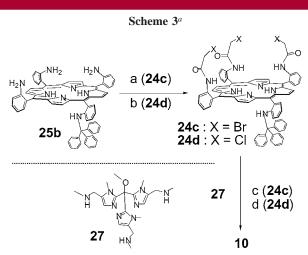
<sup>a</sup> Reagents and conditions: (a) bromoacetyl bromide, THF,  $Et_2NC_6H_4$ , 0 °C, 2 min, 86%; (b) chloroacetyl chloride,  $CH_2Cl_2$ , 15 min, rt, 80%; (c) **16**, THF,  $Et_2NC_6H_4$ , rt, 36 h (45% (**26a**)) or 4 weeks (14% (**26b**)), or 60 °C (25% (**26b**)); (d) THF,  $Et_2NC_6H_4$ , rt, 36 h, **22** (40% (**8**)) or **23** (35% (**9**)); (e) NaI, acetone, 50 °C, 16 h, then step d, 8%; (f)  $BBr_3$ ,  $CH_2Cl_2$ , -78 °C, 30 min, then 0 °C, 1 h, 40%.

imidazolic nitrogen on the bromomethylene of unreacted **24a** as pointed out earlier with imidazol acyl chlorides.<sup>3b</sup>

As expected, a gain in reactivity is obtained with TBrPP. Reaction times were slightly shorter than with TClPP, but no heating was required, which allowed recovery of the starting porphyrin; also, the yields of products were slightly increased. TBrPP is not as moisture sensitive as the very reactive isocyanatoporphyrin 4: this strategy represents a compromise between sufficient reactivity and convenience of handling. Unreacted 24a did not suffer from rotation or hydrolysis during the reaction and workup, which is normally an issue when dealing with valuable porphyrin synthons such as 26a in the synthesis of 8a or 9. TBrPP 24a is stable when stored at 4 °C, whereas dibromoacetamidopyridine-strapped porphyrin 26a was slowly converted into several unidentified porphyrins of higher polarity. Similarly, the preparation of  $\alpha_3$ [bromoacetamido]- $\beta$ -[trityl]-porphyrin  $\alpha_3$ Br $\beta$ Tr **24c**(Scheme 3) from  $\alpha_3$ [amino]- $\beta$ -[trityl]-porphyrin **25b**<sup>3f</sup> was achieved in 55% yield. A 40-fold excess of base was required to keep trityl cleavage to a minimum.  $\alpha_3 Br \beta Tr$  Synthon 24c was stable at room temperature in solution for hours. Reaction of tripodal ligand tris-imidazole-tris-secondary amine 278f with 24c led to  $\alpha\alpha\alpha\beta$ -model 10 in 28% yield together with several unidentified side products. When starting from 24d, prior halogen exchange was required and the isolated yield of 10 was only 6.5%.

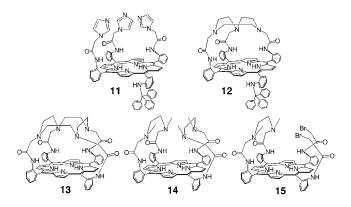
Previously reported condensations were also run with the new bromoacetamido linker. Tris-imidazole-picket- $\beta$ -trityl-

Org. Lett., Vol. 6, No. 6, 2004



<sup>a</sup> Reagents and conditions: (a) bromoacetyl bromide, THF,  $Et_2NC_6H_4$ , 0 °C, 2 min, 86%; (b) chloroacetyl chloride,  $CH_2Cl_2$ , 15 min, rt, 70%; (c) 27, THF,  $Et_2NC_6H_4$ , rt, 36 h, 28%; (d) NaI, acetone, 50 °C, 16 h, then step c, 6.5%.

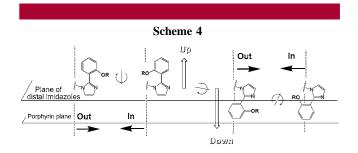
porphyrin 11<sup>3f</sup> was obtained in 40% yields by reaction of 24c with cyanoethyl-protected imidazole<sup>3f</sup> followed by treatment with sodium methylate. Capping of  $\alpha_{3^-}$  and  $\alpha_{4^-}$  porphyrins  $\alpha_3 \text{Br} \beta \text{Tr}$  24c and  $\alpha \alpha \alpha \alpha$ -TBrPP 24e with triazacyclononane and cyclen led to porphyrins 12 and 13 in 43–50% yield.



Also, mono- and bis-strapped  $\alpha_4$ -porphyrins **14** and **15**, respectively, were obtained by reaction of **24e** with dimethylethylenediamine. Contrary to **24c** and **26a**, **15** is less stable in solution, as unidentified polar fractions appeared after several hours, presumably from nucleophilic attack of the poorly hindered tertiary amine on the bromomethylene picket. Also, in capped and strapped porphyrins **12–15**, there is only one methylene per acetamido picket in contrast to previously capped and strapped porphyrins, where two methylenes per acetamido picket are present.  $^{3d-f}$  This is an issue when considering Fe/Cu distances in the metalated models.

The tris-imidazole strapped and capped models 8-10 are more polar than tris-imidazole picket porphyrins. With respect to chloroacetamido-type porphyrins, the new bro-moacetamidoporphyrins display the following physical and

spectroscopic characteristics: lower solubility in common organic solvents, smaller  $R_f$  values, and halogeno-methylene  $^1$ H NMR signals downfield-shifted (by 0.24 ppm) with bromine. 8a has theoretically several isomers: the phenoxyimidazole moiety can rotate along the C-N axis giving two possible isomers, one with the moiety up, the other with the moiety down. The phenoxy moiety can also rotate along the C(imidazole)-C(phenyl) bond: two extreme positions might be considered, inside (in) or outside (out) together with other intermediate positions. This is another kind of rotational isomer taking place on porphyrin at the superstructure and not at the meso-phenyl rings.



Interconversion was slow enough at room temperature to attempt chromatographic separation, but because of the similar polarity of the atropisomers, only one atropisomer could be purified. As for the other fractions, the same peak at m/z 1464.5 was found, and broad <sup>1</sup>H NMR peaks were obtained. Interconversion was much faster above 30–40 °C, which is reminiscent of the rotation of *meso*-amino pickets previously reported,<sup>3a</sup> which did not allow the use of high-temperature NMR with **8a,b**. With the more polar compound **8b**, obtained by BBr<sub>3</sub> treatment of **8a**, isomers were not even detected on TLC.

In conclusion, a new generation of models is reported where capped and strapped tris-imidazole porphyrins offer greater stability and no rotation of distal Cu complexes. One of them is the closest structural analogue yet reported of the metal free CcO. The bromoacetamido linker is efficient in attaching a variety of distal tris-imidazole superstructures. It offers a compromise between high reactivity and convenience in handling.

**Acknowledgment.** This material is based upon work supported by the National Institutes of Health under Grant No. 017880. Support from NSF Grant CHE-013206 also contributed to this work. Dr. Chi Zhang is acknowledged for preparing some **16** from **18**, and Dr. Minh Zhong for providing *p*-toluenesulfinic acid. R.A.D. thanks the French Foreign Ministry for a *Lavoisier* fellowship. We also thank the Mass Spectrometry Facility, University of California, San Francisco, and the Stanford University Mass Spectrometry.

Supporting Information Available: Experimental procedures and characterization data for new compounds 8–10, 12, 14–16, 19, 23–24, and 26a,b. This material is available free of charge via the Internet at http://pubs.acs.org.

OL049912M

1036 Org. Lett., Vol. 6, No. 6, 2004